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# Identification of sensitive and tolerant organisms from a tropical estuarine lagoon to heavy metal intoxication

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ABSTRACT: Sequel to the establishment of the most prominent heavy metals in industrial effluents, drainage channels and the Lagos lagoon environment, the relative susceptibility of six animals from this tropical estuarine complex to some of the identified heavy metal species was investigated using standard semi-static bio-assay techniques.

Against the test metals (Fe, Mn, Cu and Hg), *Mugil sp., T. guineensis* and *Cypris sp.,* had relatively low susceptibility thresholds and were identified as sensitive species based on results of semi-static laboratory tests.

On the other hand, *C. africanus, N. senegalensis* and *T. fuscatus* were identified as relatively tolerant animals from the Lagos lagoon system. The results are discussed in terms of their relevance to environmental management, associated standards, guidelines and consumer safety.

Key words: Coastal lagoon; Heavy metal; Sensitivity; Tolerance; Environmental management; Consumer safety.

#### Introduction

Population pressure and the preferential use of coastal space for industrialization often result in the deterioration in quality of coastal ecosystems and the vitiation of their value as multiple-usage amenities. Anthropogenic activities and influences in general are responsible for the contamination and or pollution of coastal ecosystems by critical pollutants including the class known as heavy metals.

Environmental protective measures and standards can be usefully informed by the results of field and laboratory studies including bio – monitoring and the identification of sensitive and tolerant organisms.

In West Africa, the Lagos lagoon complex is the largest of the four lagoon systems of the Gulf of Guinea coast (Hill and Webb, 1958) and probably typifies similar tropical lagoons world-wide. This lagoon system borders the rain forest belt and receives a number of major rivers and streams including Majidun, Ogun, Ona, Shasha and Oshun which drain about 103,637 km<sup>2</sup> of vast country. This expanse of water is generally shallow with a depth of between 0.3 and 3.2m in most parts with the exception of some dredged parts, notably in the Lagos harbour, where depths greater than 10m have been recorded. The tidal range is small being only 0.3-1.3m. The tides are the semi-diurnal type. The main body of this lagoon

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system lies between longitudes 3° 22' and 3° 40' East and latitudes 6° 17' and 6° 28' North. The Lagos lagoon complex also receives an input of sea water from the Atlantic Ocean into which it opens via the Lagos Harbour. The cosmopolitan city of Lagos which is built around the Lagos lagoon, may have more than 12 million inhabitants while more than 80% of all industries in Nigeria are situated in and around the metropolis. Therefore, the Lagos lagoon complex receives waste waters/effluents from industrial and domestic sources as well as urban storm waters and contaminated stream or river waters.

In order to protect this and similar ecosystems from the known and potential negative impacts of pollution by heavy metals and other critical pollutants, there have been several studies aimed at understanding relevant characteristics of the environment and man-made inputs as benchmarks for judicious management in different parts of the world (Ayodele *et al.*, 1991; Bryan and Langston, 1992; Chen *et al.*, 1991; Clements *et al.*, 1990; Flos *et al.*, 1987; Ortego and Benson, 1992). The focus of most of these studies is the determination of ambient concentrations of critical contaminants and pollutants and sometimes aspects of their biological effects. Such attempts have usually precluded ranking of the investigated animal species in order of susceptibility as a means of identifying sensitive and tolerant ones.

The plants and animals of this system have been the subject of scientific studies by previous workers. For example, Hill and Webb, (1958); Fagade (1969); Olaniyan (1969); Fagade and Olaniyan (1974); Mah-Essiet (1986); Omoniwa (1988); Ajao (1989); Ajao and Fagade (1990); Nwankwo (1996) and a body of knowledge already exists on the biology and ecology of a number of pelagic, benthic and sessile Lagos lagoon animals including *Mugil sp., Tilapia guineensis, Cypris spp, Tympanotonus fuscatus, Clibanarius africanus* and *Nerita senegalensis* which are the test animals in this study. There is however little or no information on their relative sensitivity or tolerance to contaminants and pollutants. Yet such information will be invaluable in identifying organisms that could be targeted for particular purposes in schemes designed to protect and manage this valuable ecosystem.

In an earlier study, the most prominent heavy metal species discharged by industries around the Lagos lagoon complex were identified while the magnitude of their occurrence in the environment was determined (Oyewo, 1998). The present study describes the susceptibility ranking of *Mugil spp;, Ti guineensis, Cypris spp., T. fuscatus; Clibanarius africanus* and *N. senegalensis* to some of these heavy metal species (Fe, Mn, Cu and Hg) in an attempt to identify sensitive and tolerant ones to heavy metal intoxication.

#### **Materials and Methods**

Evaluation of sensitivity/tolerance was based on laboratory bio-assays involving the test animals and metals.

#### Semi-Static Bioassay Technique

All bio-assays followed standard semi-static procedures with the observance of all necessary traditions and precautions (Sprague, 1973, Ward and Parish, 1982). Appropriate modifications were made where necessary (Oyewo, 1998; Tokolo, 1988, Alegbeleye *et al.*, 1989).

#### Test Animals

All the test animals were collected from the Ikoyi Experimental Fish Farm of the Nigerian Institute for Oceanography and Marine research which was protected, with only limited and controlled connection with the open Lagos Lagoon System. The animals were mainly from cultured stocks of known history which are usually preferred for toxicological bioassays ratherthan animals collected from the wild which may have acquired increased tolerance to pollutants over years of exposure (Callahan and Weiss, 1983; Ward and parish, 1982; Le blanc, 1982; Salami, 1990).

The test animals employed in this study included fingerlings of the mullet *Mugil sp.* (Pisces, Mugilidae) (TL =  $70 \pm 5$ mm), *Tilapia guineensis* (Pisces: Cichlidae) (Total Length (TL) =  $65 \pm 4$ mm), *Cypris sp.* (Ostracoda) (0.25 – 0.30mm diameter), *Tympanotonus fuscatus* (Mollusca; Gastropoda) (shell length  $43 \pm 4$ mm); *Clibanarius africanus* (Arthropoda, Crustacea) (weight of shucked animal =  $0.48 \pm 0.05$ g), and

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*Nerita senegalensis* (Mollusca, Gastropoda) (Length =  $7.5 \pm 2$ mm). They were subsequently transported to the laboratory and acclimatized before use in bioassays.

#### Test Temperature and Salinity

All test animals were acclimatized to a temperature regime of  $22 - 24.5^{\circ}$ C and salinity of 15 psu. The only exception was *Cypris*. It was tested under the same temperature regime but at 5 psu based on results of earlier salinity tolerance tests (Oyewo, 1998).

#### Test Chemicals

The heavy metals and salts used in this study were: Iron as FeCl<sub>2</sub>; manganese as MnSO<sub>4</sub>; Copper as CuSO<sub>4</sub>.5H<sub>2</sub>O and Mercury as HgCl<sub>2</sub>. All were of analar grade quality. With the exception of Mercury, the choice of heavy metals for this study was based on the commonest metals from the results of the chemical survey of industrial effluents, lagoon waters and sediments for heavy metals in an earlier related study (Oyewo, 1998).

#### Assessment of Quantal Response

The quantal responses adopted in this study included death, cessation of opercular or other body movement, failure to respond to prodding, failure to protrude foot during a defined observation period in untreated dilution water and loss of hold-fast capacity depending on the test animal (Oyewo, 1998).

Mortality and other responses were assessed every 24 hours.

#### **Statistics**

Toxicological dose-response data involving quantal responses were analysed by probit analysis(Finney, 1971) based on a computer programme written by Ge Le Pattourel, Imperial College, London as adopted by Don-Pedro (1996b).

#### Results

#### Iron (Fe)

Mugill *sp.* was the most susceptible test animal to fe with a response level based on 96-hr LC<sub>50</sub> value of 50.32 mg L<sup>-1</sup> followed by *Cypris sp., Ti. Guineensis, N. senegalensis* and *T. fuscanus* (the most tolerant species, 96-hr LC<sub>50</sub> = 17409 mg L<sup>-1</sup>) in a descending order of susceptibility (Tables 1 – 6). The susceptibility thresholds of the three most sensitive test animals, *Mugi, Tilapia* and *Cypris sp.,* to Fe were significantly lower than those of the more tolerant animals, *N. senegalensis* and *T. fuscatus* by several orders of magnitude. *C. africanus* was not tested against Fe.

#### Manganese (Mn)

*Cypris sp.* was the most susceptible to Mn with a response level, based on 96-hr LC<sub>50</sub> value of 90.14 mg L<sup>-1</sup> followed by *Mugil sp., T. guineensis, N. senegalensis, C, africanus* and *T. fuscatus* (the most tolerant species, 96-hr LC<sub>50</sub> = 8409 mg L<sup>-1</sup>) in a descending order of susceptibility (Tables 1 – 6). The susceptibility threshold of the three most sensitive test animals; *Mugil sp., Tilapia* and *Cypris* were significantly (no overlaps in 95% CL of 96-hr LC<sub>50</sub> values) lower than those of the more tolerant animals – *N. senegalensis, C. africanus* and *T. fuscatus* (Table 1-6).

Time (h)	(LC <sub>50</sub> mg/L} 95% Confidence Limits	LC <sub>95</sub> (mg/L} 95% Confidence Limits	LC <sub>5</sub> (mg/L} 95% Confidence Limits	Chi sqd	Df	Р	Slope $\pm$ SE	Probit line Equation			
IRON (Fecl <sub>2</sub> )											
24	137.20(104.49-179.94)	655.73(408.62-1049.37)	28.71(15.92-51.78)	5.518	4	0.238	2.43 <u>+</u> 0.36	Y = -0.191 + 2.43x			
48	71.71(54.24-94.68)	275.52(177.83-426.93)	18.67(10.98-31.65)	1.201	3	0.753	$2.82 \pm 0.43$	Y = -0.237 + 2.82x			
72	66.82(50.94-87.62)	227.79(142.65-364.35)	19.60(11.92-32.16)	1.424	2	0.491	3.10 <u>+</u> 0.51	Y = -0.653 + 3.10x			
96	50.32(37.92-66.71)	202.04(113.56-360.52)	12.54(6.37-24.55)	1.255	2	0.534	2.73 <u>+</u> 0.56	Y = 0.349 + 2.73x			
	MANGANESE (MnSo4)										
24	209.70(157.71-278.51)	1096.04(665.6-1799.3)	40.12(21.86-73-70)	2.284	4	0.684	2.30+0.34	Y = -0.333 + 2.30x			
48	129.99(99.41-169.78)	530.58(335.93-835.71)	31.85(17.71-57.29)	3.298	3	0.348	2.70 <u>+</u> 0.44	Y = -0.710 + 2.70x			
72	121.03(95.61-153.12)	451.81(283.51-717.65)	32.42(18.23-57.77)	0.384	3	0.943	2.88 <u>+</u> 0.52	Y = -1.008 + 2.88x			
96	121.03(95.61-153.12)	451.81(283.51-717.65)	32.42(18.23-57.77)	0.384	3	0.943	2.88 <u>+</u> 0.52	Y = -1.008 + 2.88x			
			COPPER (CuS <sub>04</sub> .5I	H <sub>20</sub> )							
24	0.37(0.30-0.45)	1.48(1.09-2.14)	0.09(0.06-0.14)	8.623	4	0.071	2.72 <u>+</u> 0.33	Y = 6.187 + 2.72x			
48	0.20(0.16-0.25)	0.72(0.49-1.09)	0.05(0.03-0.09)	2.696	2	0.260	2.95 <u>+</u> 0.44	Y = 7.078 + 2.95x			
72	0.19(0.15-0.23)	0.59(0.41-0.86)	0.06(0.04-0.09)	1.989	2	0.370	3.33 <u>+</u> 0.50	Y = 7.423 + 3.33x			
96	0.16(0.12-0.19)	0.55(0.36-0.83)	0.04(0.03-008)	2.902	2	0.234	3.02 <u>+</u> 0.54	Y = 7.448 + 3.02x			
	MERCURY (Hgcl2)										
24	0.053(0.042-0.067)	0.25(0.17-0.37)	0.011(0.007-0.018)	6.34	4	0.175	2.42 <u>+</u> 0.28	Y = 8.094 + 2.42x			
48	0.031(0.025-0.039)	0.12(0.09-0.174)	0.008(0.005-0.013)	5.18	3	0.159	2.80+0.36	Y = 9.201 + 2.80x			
72	0.020(0.016-0.024)	0.07(0.05-0.103)	0.006(0.003-0.009)	2.74	3	0.433	2.97 + 0.44	Y = 10.040 + 2.97x			
96	0.018(0.015-0.023)	0.07(0.05-0.102)	0.005(0.003-0.008)	2.97	3	0.396	2.81 <u>+</u> 0.43	Y = 9.889 + 2.81x			

# TABLE 1: THE RELATIVE SUSCEPTIBILITY OF *Mugil* sp TO SELECTED HEAVY METALS.

Time (h)	(LC <sub>50</sub> mg/L} 95% Confidence Limits	LC <sub>95</sub> (mg/L} 95% Confidence Limits	LC <sub>5</sub> (mg/L} 95% Confidence Limits	Chi sqd	df	Р	Slope $\pm$ SE	Probit line Equation
			IRON (Fecl <sub>2</sub> )					
24	476.90(393.28-578.07)	2064.24(1475.5-2867.7)	110.18(71.02-171.98)	7.001	5	0.221.	2.59 <u>+</u> 0.31	Y = -1945 + 2.59x
48	314.20(254.14-388.21)	1322.93(946.77-1836.9)	74.62(47.72-117.28)	7.342	4	0.119	2.64 <u>+</u> 0.31	Y = -1.599 + 2.64x
72	314.15(259.43-380.30)	1320.99(902.21-1922.4))	74.71(48.42-115.91)	5.133	4	0.274	$2.65 \pm 0.34$	Y = -1.605 + 2.65x
96	239.65(195.99-292.88)	992.33(695.75-1406.69)	57.88(36.28-92.79)	3.895	4	0.420	2.6 <u>7</u> +0.35	Y = -1.362 + 2.67x
			MANGANESE (Mn	So4)				
24	976.17(768.12-1241.12)	5133.3(2780-9423)	185.6(112.1-309.4)	1.742	3	0.767	2.29+0.36	Y = -1.842 + 2.29x
48	470.98(384.83-576.13)	1844.4(1333-2533)	120.27(78.15-186.09)	3.601	4	0.463	2.78 + 0.34	Y = -2.439 + 2.78x
72	292.88(232.54-368.65)	1117.1(819.57-1670.9)	73.12(47.39-113.22)	8.231	3	0.041	$2.74 \pm 0.33$	Y = -1.753 + 2.74x
96	231.39(185.14-289.01)	920.22(641.4-1313.4)	58.19(36.60-92.85)	3.250	3	0.355	$2.75 \pm 0.36$	Y = -1.507 + 2.75x
			COPPER (CuS <sub>04</sub> .51	H <sub>20</sub> )				
24	0.44(0.33-0.56)	1.70(1.13-2.93)	0.11(0.06-0.18)	7.206	3	0.066	2.79 <u>+</u> 0.43	Y = 6.008 + 2.79x
48	0.29(0.21-0.39)	1.53(0.92-2.89)	0.06(0.03-0.10)	4.407	3	0.221	$2.28 \pm 0.37$	Y = 6.231 + 2.28x
72	0.19(0.14-0.27)	0.97(0.60-1.74)	0.04(0.02-0.08)	1.746	3	0.627	$2.35 \pm 0.41$	Y = 6.677 + 2.35x
96	0.16(0.11-0.22)	0.71(0.41-1.34)	0.03(0.01-0.08)	2.257	2	0.323	2.50 <u>+</u> 0.52	Y = 7.016 + 2.50x
			MERCURY (Hgc	l <sub>2</sub> )				
24	0.10(0.08-0.13)	0.74(0.47-1.19)	0.014(0.008-0.024)	10.49	4	0.062	1.91+0.22	Y = 6.904 + 1.91x
48	0.07(0.05-0.09)	0.49(0.30-0.82)	0.010(0.005-0.018)	7,90	4	0.096	$1.94 \pm 0.24$	Y = 7.246 + 1.94x
72	0.03(0.02-0.04)	0.14(0.088-0.212)	0.006(0.003-0.010)	2.40	3	0.504	2.40+0.32	Y = 8.724 + 2.40x
96	0.02(0.017-0.03)	0.095(0.061-0.146)	0.005(0.003-0.008)	3.20	3	0.362	$2.51 \pm 0.38$	Y = 9.224 + 2.51x

# TABLE 2: THE RELATIVE SUSCEPTIBILITY OF Tilapia guineensis TO SELECTED HEAVY METALS.

Time (h)	(LC <sub>50</sub> mg/L} 95% Confidence Limits	LC <sub>95</sub> (mg/L} 95% Confidence Limits	LC <sub>5</sub> (mg/L} 95% Confidence Limits	Chi sqd	df	Р	Slope <u>+</u> SE	Probit line Equation			
	IRON (Fec/2)										
24 48 72 96	21064(177.99-249.20) 201.88(170.18-239.39) 175.80(138.61-222.98) 90.08(68.62-118.23)	786.23(574.69-1069.10) 758.78(556.28-1028.69) 1379.11(763.27-2481.80) 684.76(344.72-1361.86)	56.43(38.28-83.65) 53.71(36.31-79.88) 22.41(12.47-40.45) 11.85(5.76-24.33)	10.523 8.252 7.194 4.168	5 5 5 3	0.062 0.143 0.207 0.244	2.88 <u>+</u> 0.35 2.87 <u>+</u> 0.34 1.84 <u>+</u> 0.25 1.87+0.30	Y = -1.702 + 2.88x Y = -1.614 + 2.87x Y = 0.859 + 1.84x Y = 1.339 + 1.87x			
	MANGANESE (MnSo4)										
24 48 72 96	293.33(249.90-344.22) 249.06(211.10-293.78) 154.48(121.13-196.85) 90.14(68.02-119.24)	908.27(687.52-1193.11) 939.01(669.73-1308.63) 1112.01(693.11-1778.27) 709.97(409.60-1228.18)	94.73(65.53-137.66) 66.06(45.12-97.25) 21.44(12.35-37.36) 11.45(5.70-22.94)	10.282 9.695 16.521 10.412	4 5 5 4	0.036 0.084 0.006 0.034	3.36 <u>+</u> 1.43 2.86 <u>+</u> 0.35 1.92 <u>+</u> 0.23 1.84 <u>+</u> 0.25	Y = -3.293 + 3.36x Y = -1.860 + 2.86x Y = 0.789 + 1.92x Y = 1.401 + 1.84x			
			COPPER (CuS <sub>04</sub> .5H	( <sub>20</sub> )							
24 48 72 96	0.32(0.28-0.38) 0.25(0.22-0.29) 0.22(0.19-0.26) 0.16(0.14-0.19)	1.47(1.11-1.99) 1.14(0.88-1.52) 1.09(0.79-1.55) 0.77(0.55-1.08)	0.07(0.05-0.09) 0.06(0.04-0.07) 0.04(0.03-0.06) 0.04(0.02-005)	12.504 8.120 10.885 3.215	6 7 6 5	0.052 0.322 0.092 0.667	2.51 <u>+</u> 0.23 2.52 <u>+</u> 0.20 2.36 <u>+</u> 0.21 2.46 <u>+</u> 0.24	Y = 6.232 + 3.36x Y = 6.503 + 2.52x Y = 6.558 + 2.36x Y = 6.935 + 2.46x			
	MERCURY (Hgcl <sub>2</sub> )										
24 48 72 96	0.30(0.22-0.38) 0.16(0.12-0.21) 0.13(0.10-0.16) 0.12(0.10-0.15)	1.60(1.102-2.496) 0.86(0.55-1.39) 0.84(0.53-1.39) 0.61(0.39-0.99)	0.054(0.030-0.088) 0.029(0.017-0.047) 0.019(0.010-0.033) 0.022(0.013-0.037)	6.453 2.925 3.958 5.905	4 3 4 3	0.168 0.403 0.412 0.116	2.25 <u>+</u> 0.26 2.24 <u>+</u> 0.26 2.00 <u>+</u> 0.25 2.30 <u>+</u> 0.31	$\begin{split} Y &= 6.193 + 2.25x \\ Y &= 6.801 + 2.24x \\ Y &= 6.803 + 2.00x \\ Y &= 7.143 + 2.30x \end{split}$			

## TABLE 3: THE RELATIVE SUSCEPTIBILITY OF *Cypris* sp. TO SELECTED HEAVY METALS.

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#### Copper (Cu)

*T. guineensis, Mugil sp. and Cypris sp.* had similar magnitude of responses to Cu, (with overlaps in 95% Cl of 96-hr  $LC_{50}$  values) and therefore constituted the three most susceptible animal species to Cu with response levels of approximately 0.16 mg L<sup>-1</sup> in each case, followed by *C. africanus, N. senegalensis* and *T. fuscatus* (the most tolerant sp., 96-hr  $LC_{50} = 8.84 \text{ mg L}^{-1}$ ) in a descending order of susceptibility (tables 1-6). The response level of any of the three most sensitive animal species was significantly (no overlaps in 955% Cl of 96-hr  $LC_{50}$  values) higher than those recorded for the most tolerant animals – *C. africanus, N. senegalensis* and *T. fuscatus*.

#### Mercury (Hg)

*Mugil sp.* was the most susceptible to Hg with a response level, based on 96-hr LC<sub>50</sub> value of 0.018 mg L<sup>-1</sup> followed by *T. guineensis, Cypris sp., C. africanus, N. senegalensis* and *T. fuscatus* (the most tolerant sp. 96-hr LC<sub>50</sub> = 2.47mg L<sup>-1</sup>) in a descending order of susceptibility (Tables 1-6). The susceptibility threshold of the three most sensitive test animals; *Mugil sp., T. guineensis* and *Cypris* were significantly lower than those recorded for the more tolerant test animals – *C. africanus, N. senegalensis* and *T. fiscatus*.

#### Discussion

The series of acute bioassays carried out permitted the establishment of a susceptibility scale for six resident Lagos lagoon animal species, against each test metal. Overall, *Mugil sp., T. guinensis* and *Cypris sp.* were the most susceptible and were significantly less tolerant to the heavy metals than the three other test species namely; *N. senegalensis, C. africanus* and *T. fuscatus* – the least susceptible sp. in all cases. Although it generally recognized that results of laboratory tests can not be directly extrapolated to predict field toxicity situations accurately, there is a regular attempt to correlate results of laboratory toxicity tests with field toxicity in order to explain observations in the wild (baron, 1995). In spite of their limitations, result of laboratory test are known to serve as fair indications of field situations and have been variously used to understand complex field interactions or complement field studies. Thus, the test animals can be broadly classified as sensitive and tolerant species based on the results obtained. It is however desirable to carry out similar screening on other organisms and against other critical pollutants to enrich the data pool for appropriate and informed decisions.

The differential susceptibility shown by the test animals to heavy metals serves as an extension of scientific knowledge that has been demonstrated elsewhere (Torres *et al.*, 1987; Mackie, 1989; Chen *et al.*, 1991). In general, differential responses amongst organisms have been attributed to factors such as; nature of the cuticle or body covering with respect to penetrability, metabolic transformation capacities of the organisms; for example, availability of the right type of enzymes and optimal physico-chemical conditions, excretory capacity and the rate of elimination of the by-products of metabolism, availability and sensitivity of site of action, body size, age and life cycle stage as well as ecology with particular reference to location and activity coefficient and possibly, behavioural attributes (Don-Pedro, 1987; Don-Pedro, 1996b). The observed differential susceptibilities are thus expected.

The susceptibility or sensitivity scale established in this work represents useful information which in conjunction with similar ones will be invaluable for the management of the Lagos lagoon and similar ecotypes with respect to heavy metal pollution in a number of ways.

The most sensitive species can serve as sensitive indicators that can be employed in the early detection of heavy metal pollution since they respond at low contamination levels thereby serving as early warning signals.

More efficient and wider environmental protection of living organisms can be achieved by employing the response levels of the more sensitive species as a basis or starting point from which no effect levels and consequently safe limits can be extrapolated and employed in fixing realistic standards and or allowable levels in effluents and the recipient natural ecosystems such as the Lagos lagoon.

Time (h)	(LC <sub>50</sub> mg/L} 95% Confidence Limits	LC <sub>95</sub> (mg/L} 95% Confidence Limits	LC <sub>5</sub> (mg/L} 95% Confidence Limits	Chi sqd	Df	Р	Slope + SE	Probit line Equation		
			IRON (Fecl <sub>2</sub> )							
24 48 72 96	28788(1879544104) 35930(19120-67584) 20989(16898-26075) 17409(15040-20153)	59335(22631-154967) 140691(31728-620193) 58153(31556-106560) 42353(27990-63765)	13967(11413-17166) 9176(6419.1-13219.94) 7575(5777-9993) 7156(5544-9284)	0.047 0.320 0.223 2.567	1 2 3 2	0.829 0.852 0.974 0.277	5.25 <u>+</u> 0.04 2.783 <u>+</u> 0.92 3.73 <u>+</u> 0.78 4.27 <u>+</u> 0.76	$\begin{array}{l} Y = -1.8.424 + 5.25x \\ Y = -7.679 + 2.78x \\ Y = -11.112 + 3.73x \\ Y = -13.122 + 4.27x \end{array}$		
			MANGANESE (Mn	So4)						
24 48 72 96	28915(11823-70786) 19154(15483-23699) 14011(12612-15566) 8409(7273-9722)	145167(6310.01-3318989) 51087(27822-93312) 29763(22061-39980) 24659(18471-32722)	5759.7(1365.4-244946) 7181.5(5169.4-10032.8) 6596.2(5173.11-8447.9) 2867.6(2073.7-3989.0) COPPER (CuSO4.5]	0.593 0.734 1.224 4.294 H20)	1 2 3 4	0.441 0.693 0.747 0.368	$\begin{array}{c} 2.87 \pm 1.69 \\ 3.87 \pm 0.89 \\ 5.04 \pm 0.88 \\ 3.53 \pm 0.45 \end{array}$	Y = -5.504 + 2.355x Y = -11.583 + 3.87x Y = -15.909 + 5.04x Y = -8.859 + 3.53x		
24 48 72 96	19.24(15.04-24.60) 13.26(10.76-16.33) 11.66(9.53-14.26) 8.84(7.05-11.09)	133.36(83.54-214.9) 64.52(44.31-94.46) 48.93(33.46-72.04) 43.84(27.03-72.41)	2.78(1.68-4.53) 2.73(1.77-4.17) 2.78(1.84-4.17) 1.78(1.05-2.97)	4.876 2.138 3.788 7.722	5 5 4 3	0.431 0.830 0.435 0.052	1.96±0.21 2.40±0.27 2.65±0.32 2.37±0.34	$\begin{split} Y &= 2.480 + 1.96x \\ Y &= 2.305 + 2.40x \\ Y &= 2.174 + 2.65x \\ Y &= 2.754 + 2.37x \end{split}$		
	MERCURY (Hgcl2)									
24 48 72 96	5.09(4.42-5.87) 3.49(3.16-3.87) 2.68(2.50-2.87) 2.47(2.31-2.65)	17.16(12.98-22.84) 7.79(6.20-9.85) 4.57(3.95-5.29) 4.01(3.55-4.55)	1.51(1.12-2.04) 1.57(1.25-1.95) 1.57(1.34-1.84) 1.52(1.31-1.76)	3.129 4.336 3.107 5.211	6 5 5 4	$\begin{array}{c} 0.762 \\ 0.502 \\ 0.684 \\ 0.266 \end{array}$	3.13 <u>+</u> 0.33 4.74 <u>+</u> 0.61 7.11 <u>+</u> 0.91 7.84 <u>+</u> 0.97	$\begin{split} Y &= 2.788 + 3.13x \\ Y &= 2.426 + 4.74x \\ Y &= 1.956 + 7.11x \\ Y &= 1.918 + 7.84x \end{split}$		

### TABLE 4: THE RELATIVE SUSCEPTIBILITY OF Tympanotonus fuscatus TO SELECTED HEAVY METALS.

Time (h)	(LC <sub>50</sub> mg/L} 95% Confidence Limits	LC <sub>95</sub> (mg/L} 95% Confidence Limits	LC <sub>5</sub> (mg/L} 95% Confidence Limits	Chi sqd	df	Р	Slope + SE	Probit line Equation			
	MANGANESE (MnS04)										
24 48 72 96	36109(14701-89390) 18344(10710-31504) 12563(8467-18658) 5057(3515-7270)	1338548(65970-27452909) 405685(59988-2740839) 217397(54913-854786) 83155(36883-185914)	974.1(191.12-4988) 829.6(211.3-3277.9) 726.0(236.2-2251.6) 307.6(124.5-765.7)	4.672 11.065 10.521 9.870	2 2 3 4	0.097 0.004 0.015 0.043	1.05 <u>+</u> 0.33 1.233 <u>+</u> 0.32 1.33 <u>+</u> 0.28 1.36 <u>+</u> 0.19	$\begin{array}{lll} Y = & 0.207 + 1.05 x \\ Y = & -0.231 + 1.23 x \\ Y = & -0.462 + 1.33 x \\ Y = & -0.026 + 1.36 x \end{array}$			
	COPPER (CuSo4.5H20)										
24 48 72 96	36.96(29.83-45.76) 0.59(0.52-0.66) 0.40(0.36-0.45) 0.38(0.34-0.43)	160.34(105.94-242.5) 1.42(1.13-1.86) 0.89(0.73-1.11) 0.83(0.68-1.01)	8.52(5.32-13.62) 0.24(0.19-0.30) 0.18(0.14-0.23) 0.18(0.14-0.22)	8.26 1.97 6.02 1.16	3 5 4 4	0.041 0.854 0.198 0.884	2.59 <u>+</u> 0.32 4.31 <u>+</u> 0.52 4.73 <u>+</u> 0.58 4.97 <u>+</u> 0.61	Y = 0.942 + 2.59x Y = 5.991 + 4.31x Y = 6.889 + 4.73x Y = -7.065 + 4.97x			
	MERCURY(Hgcl2)										
24 48 72 96	43.42(36.08-52.22) 2.40(1.93-2.98) 0.40(0.36-0.45) 0.38(0.34-0.43)	148.37(102.46-214.14) 10.22(7.16-15.49) 0.89(0.73-1.11) 0.89(0.71-1.13)	12.71(8.003-20.22) 0.57(0.36-0.83) 0.18(0.14-0.23) 0.17(0.12-0.22)	8.91 2.19 6.02 9.28	3 4 4 3	0.031 0.701 0.198 0.026	3.09±0.48 2.62±0.31 4.73±0.58 4.56±0.63	$\begin{split} Y &= 0.064 + 3.09x \\ Y &= 4.001 + 2.62x \\ Y &= 6.889 + 4.73x \\ Y &= 6.891 + 4.56x \end{split}$			

## TABLE 5: THE RELATIVE SUSCEPTIBILITY OF *Clibanarius africanus* TO SELECTED HEAVY METALS.

Time (h)	(LC <sub>50</sub> mg/L} 95% Confidence Limits	LC <sub>95</sub> (mg/L} 95% Confidence Limits	LC <sub>5</sub> (mg/L} 95% Confidence Limits	Chi sqd	Df	Р	Slope <u>+</u> SE	Probit line Equation
			IRON (Fecl <sub>2</sub> )					
24 48 72 96	2369.88(2154.9-2606.2) 1352.08(1117.8-1551.9) 1200.79(1042.7-1382.6) 1051.14(894.99-1234.50)	5547.9(4495.6-6813.1) 3962.5(3180.3-4907.5) 3341.9(2652.5-4186.9) 3488.02(2565.1-4714.3)	1012.3(789.16-1304.97) 461.3(339.9-629.69) 431.46(318.9-586.82) 316.77(231.67-435.73)	32.082 10.762 11.070 6.648	5 4 3 3	0.00 0.029 0.001 0.084	$\begin{array}{r} 4.47 \pm 0.56 \\ 3.53 \pm 0.38 \\ 3.71 \pm 0.43 \\ 3.17 \pm 0.36 \end{array}$	$\begin{array}{lll} Y = & 10.073 + 4.47x \\ Y = & 6.062 + 3.53x \\ Y = & 6.430 + 3.71x \\ Y = & 4.571 + 3.17x \end{array}$
			MANGANESE (MnS	504)				
24 48 72 96	751.64(723.99-780.34) 612.73(573.05-655.16) 509.49(465.98-557.04) 445.87(401.71-494.84)	1021.99(949.73-1097.72) 1123.05(983.68-1277.6) 1105.08(930.91-1305.98) 1087.53(906.06-1298.73)	552.81(512.38-597.52) 334.31(282.57-396.92) 234.90(187.74-295.20) 182.80(141.9-236.57) COPPER (CuS <sub>04</sub> .5F	5.614 18.718 15.156 19.859	4 5 5 5	0.230 0.002 0.010 0.001	12.37 <u>+</u> 1.33 6.27 <u>+</u> 0.72 4.91 <u>+</u> 0.58 4.26 <u>+</u> 0.48	$\begin{array}{l} Y = -30.561 + 12.37x \\ Y = -12.477 + 6.27x \\ Y = -8.283 + 4.90x \\ Y = -6.287 + 4.261x \end{array}$
24 48 72 96	3.98(3.39-4.66) 2.87(2.40-3.42) 2.11(1.73-2.56) 1.65(1.38-1.98)	15.98(12.08-21.40) 15.98(11.59-22.52) 12.99(9.06-19.98) 7.82(5.68-11.80)	0.99(0.70-1.37) 0.52(0.36-0.73) 0.34(0.22-0.50) 0.35(0.24-0.47)	14.72 11.54 6.50 5.29	5 6 5 5	0.012 0.0.73 0.265 0.382	$\begin{array}{c} 2.73 \pm 0.27 \\ 2.22 \pm 0.19 \\ 2.09 \pm 0.20 \\ 2.44 \pm 0.23 \end{array}$	$\begin{split} Y &= 3.363 + 2.73x \\ Y &= 3.985 + 2.22x \\ Y &= 4.323 + 2.09x \\ Y &= 4.469 + 2.44x \end{split}$
			MERCURY (Hgcl	2)				
24 48 72 96	1.87(1.69-2.07) 1.36(1.23-1.50) 1.03(0.92-1.15) 0.99(0.88-1.09)	4.84(4.06-5.85) 3.41(2.85-4.25) 2.68(2.26-3.35) 2.37(1.99-2.99)	0.72(0.58-0.89) 0.54(0.43-0.66) 0.40(0.300-0.49) 0.41(0.31-0.50)	19.064 6.866 2.758 4.084	6 5 5 4	0.004 0.231 0.737 0.395	$\begin{array}{c} 4.00 \pm 0.37 \\ 4.13 + 0.41 \\ 3.98 \pm 0.40 \\ 4.34 \pm 0.47 \end{array}$	$\begin{array}{l} Y = 3.912 + 4.00x \\ Y = 4.448 + 4.13x \\ Y = 4.944 + 3.98x \\ Y = 5.027 + 4.34x \end{array}$

 TABLE 6: THE RELATIVE SUSCEPTIBILITY OF Nerita senegalensis TO SELECTED HEAVY METALS.

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The more tolerant species identified in this work namely: *N. senegalensis, C. africanus* and *T. fuscatus* could also serve as indicator species by providing information on the state of levels of metal contamination in the environment over a period of time. Thus, by measuring the concentration of metals accumulated in the bodies of these relatively tolerant species at regular intervals the state of pollution can be extrapolated and used to further control or regulate the pollutant being monitored.

The tolerance of *T. fuscatus* in particular against heavy metal intoxication has public health implications because it is a delicacy in several Nigerian communities. Even when apparently healthy, their contaminant burden can be transferred to animals like man at higher trophic levels.

#### References

- Ajao, E.A. (1989). The influence of domestic and industrial effluents on populations of sessile and benthic organisms in Lagos lagoon. Ph.D. Thesis, University of Ibadan, 411 pp.
- Ajao, E.A. and Fagade, S.O. (1990). A study of sediments and communities in Lagos Lagoon, Nigeria. Oils and Chem. Pollut., 7, 85 – 117.
- Alegbeleye, W.O.; Awa, J.; Adetayo, J. and Ugwumba, A.A. (1989). Experimental production of natural food organisms for fry and fingerlings of cultured fish species. In: pp. 39 – 40, Annual Report of the Nigerian Institute for Oceanography and MarineResearch (NIOMR) 1989. Federal Ministry of Science and Technology, Lagos, Nigeria.
- Ayodele, J.T.; Momoh, R.U. and Aminu, M. (1991). Determination of Heavy Metals in Sharada Industrial Effluents. In: Book of Abstracts; Second National Environmental Seminar, Page 14. FEPA, F.M.H., Natural Water Resources Institute, W.H.O., Kaduna State Water Board.
- Baron, M.O. (1995). Bioaccumulation and Bioconcentration in Aquatic Organisms. In: Handbook of Ecotoxicology, pp. 652 – 662 (Hoffman, D.H.; Rattner, B.A.; Burton, B.A. Jr. and Cairns, G.A. Jr. Eds.) CRC Press Inc., - Lewis Publishers, London, Tokyo.
- Bryan, G.W. and Langston, W.J. (1992). Bioavailability, accumulation and effects of heavy metals in sediments with special reference to United Kingdom estuaries. A review. Environ. Pollut. 76: 86 131.
- Callahan, P. and Weiss, J.S. (1983). Methyl mercury effects on regeneration and ecdysis in fiddler crabs (Uca pugilator and Uca pugnax) after short term and chronic pre-exposure. Arch- Environ. Contam. Toxicol. 12(6): 707 714.
- Chen, Y.; Dai, Q.; Chen, C.; Jianming, P. and Pai, Y. (1991). The toxicity of Ag<sup>+</sup> to fish and other aquatic animals. J. Fish China Scuichan Xuebao. 15(1): 55 61.
- Clements, W.H.; Cherry, D.S. and cairns, J. (Jr.) (1990). Macro-invertebrate community response to Copper in laboratory and field experimental streams. Arch. Environ. Contam.Toxicol. 19(3); 361 365.
- Don-Pedro, O.P. (1987). Differential responses of Perennial Salt-marsh Plants to Oil Pollution. Ph.D. Thesis, Imperial College of Science and Technology, University of London, 258p.
- Don-Pedro, K.N. (1996b). Fumigant Toxicity of citrus peel oils against adult and immature stages of storage insect pests. Pestic. Sci. 47: 213 223.
- Fagade, S.O. (1969). Studies on the biology, some fishes and fisheries of the Lagos lagoon. Ph.D. Thesis, University of Lagos, 358 pp.
- Fagade, S.O. and Olaniyan, C.I.O. (1974). On the biology of *Tilapia guineensis* (Dumeril) from the Lekki Lagoon. Bull d'IFAN XXXVI. Ser. A1: 245 252.

Finney, D. J. (1971) Probit Analysis. 3rd Edition. Cambridge University Press.

- Flos, R., Tort, L. and Balasch, J. (1987) Effects of zinc sulphate on haematological parameters in the dogfish Scyliorhinus canicula and influence of MS 222. Marine Environ. Res. 21, 289 – 298.
- Hill, M. B. and Webb, J. E. (1958) The ecology of Lagos Lagoon. II. The topography and physical features of Lagos harbour and Lagos Lagoon. Phil. Trans. Roy. Soc. Bull. 241, 319 333.
- Le Blanc, G. A. (1982) Laboratory investigation into the development of resistance of *Daphnia magna* (Stromss) to environmental pollutants. Environ. Pollut. (Series A) 27, 309 322.
- Mackie, G. L. (1989) Tolerance of five benthic invertebrates to hydrogen ions and metals (Cd, Pb, Al). Arch. Environ. Contam. Toxicol. 18(1-2), 215 223.
- Mah-Essiet, E. N. (1986) Production of *N. senegalensis* (Gmelin) on the rocky shore of Tarkwa Bay, Lagos. M.Sc. Thesis, University of Lagos. 68 pp.
- Nwankwo, D. I. (1996) Phytoplankton diversity and succession in Lagos Lagoon, Nigeria. Arch. Hydrobiol. 135(4), 529 542.
- Olaniyan, C. I. O. (1969) The seasonal variation in the hydography and total plankton of the Lagoons of South West Nigeria. Nig. J. Sci. 3(2), 101 119.
- Omoniwa, F. I. (1988) A study of the nutritive value of the periwinkle *Tympanotonus fuscatus* (var Radula L). M. Sc. Thesis, University of Lagos. 57 pp.

- Ortego, L. S. and Benson, W. H. (1992) Effects of dissolved humic material on the toxicity of selected pyrethroid insecticides. Environ. Toxicol. Chem. 11(2), 261 – 265.
- Oyewo, E. O. (1998) Industrial sources and distribution of heavy metals in Lagos Lagoon and their biological effects on estuarine animals. Ph.D. Thesis, University of Lagos. 279 pp.
- Salami, O. (1990) Studies on the toxicity of insecticides against populations of *Clibanarius africanus* from three subzones of Lagos Lagoon. M.Sc. Thesis, Department of Biological Sciences, University of Lagos. 54 pp.
- Sprague, J. B. (1973) 'The ABCs of Pollutant Bioassay Using Fish.' Biological Methods for the assessment of water quality. ASTM STP 528. American Society for Testing and Materials pp 6 - 30
- Tokolo, C. A. (1988) Studies on the toxicity of some Nigerian crude oils against *Tympanotonus fuscatus* var radula L. Under different salinity regimes Msc. Thesis. University of Lagos, Nigeria.
- Torres, P. Tort, L., Hernandez Pascual, M. D. & Flos, R. (1987) Effects of heavy metals on metabolism of the dogfish *Scyliorhinus canicula* Ichtyophysiol. Acta No 11 73-82.
- Ward, G. S. & Parish, P.R. (1982). Manual of methods in aquatic environmental research, Part 6. Toxicity tests. FAO Fisheries Technical Paper No. 185.